

Useful Methods for Processing N-dimensional X-ray Datacubes

A Volcanological Example



Ed Vicenzi Department of Mineral Sciences Washington, DC

Outline

 \checkmark Concept of localized bulk chemistry on the microscale

Introduction to MVS, the spatial simplicity method, and paths to chemical phase analysis

 \checkmark The tale of Parícutin Volcano

Phase results from Parícutin specimens

Microscope: X-ray detector: Analyzer: Software: FEI Nova NanoSEM600 an FE-VP SEM Si(Li)/40 mm²/ThermoFisherSci-Nanotrace NSS 300 NSS v2/Compass (AXSIA)/X Phase



Ultimate Analytical Objectives

- Measure area (volume) fraction of all chemical phases present
- Measure composition of all phases of interest (POI)
- Normalize volume fractions and weight phase microchemistry accordingly

Sounds simple! and leads to the determination of a sample's Local Bulk Chemistry (LBC)

Local Bulk Composition?

Determination of the chemical nature of a subvolume of a sample. It often yields information about the specimen's history e.g. by way of comparison with bulk properties

Traditional approaches

• Estimate area fractions by segmenting BSE image, perform representative point analyses of each POI Fraught with problems

Analyze sum raster Fahgetaboutit!
 Ideal approach

Base area estimate on segmentation of 10³-10⁴ Xray analyses. POI composition from aggregate of spectra at all locations, yielding improved statistics

Some Applications

Earth Sciences

 reconstruction of trapped liquid composition in complex melt inclusions (eg w/daughter crystals)

 crystal- melt geothermometry where glass composition is not accessible owing to groundmass crystallization (element partitioning in general)

 estimation of fluid chemistry by summing all precipitated material in a vein

Materials Sciences

heterogeneity of welds





Method I: Mining Hyperspectral Datasets in N-Dimensional Compositional Space

Traditional Factor Analysis: Classical Principal Components



The Alternate Approach of Keenan & Kotula Spatial or Spectral Simplicity Solutions

An infinite # solutions possible (rotational ambiguity) in N dimensions. We seek only those that give us insight.



Spectral simplicity*

High spectral contrast solution.

Spatial simplicity

High spatial contrast solution.

Spatial Simplicity (Keenan and Kotula, unpublished)

- $D = CS^{T}$ (Goal: Factor raw data into C and S...linear model)
 - D is an *m*-pixel × *n*-channel raw spectral-data matrix
 - S is an $n \times p$ matrix containing the p pure-component spectra shapes
 - C is an $m \times p$ matrix containing their spatial distributions/abundances
- Data is scaled to account for non-uniform (Poisson) noise*

Allows for noise on signals of all levels to be accounted for

- PCA is performed on the scaled data such the the spatial components are orthonormal
 - Instead of classical PCA (max variance in spectral domain), maximize variance in the spatial domain.
- Rotate the spatial components to maximize their mutual simplicity with the VARIMAX procedure

Rotate PCA_{sp} solution in N dimensions to obtain maximum # zeros in one dimension of the matrix.

*M.R. Keenan and P.G. Kotula, Surf. Int. Anal. 36 (2004) 203-212.

Spatial Simplicity (contin.)

Apply the inverse rotation to the spectral components

Necessary to maintain the product $[C \times S]$

Note: The VARIMAX solution fits the data as well as the PCA solution, however yields more physical insight into the specimen by virtue of the assumption that simplicity is desirable and defined as maximized contrast.

Perform non-negativity-constrained *classical* least squares (CLS) on the data, making spatial components all positive

D (known) = C (constrained) × S Hence, set all neg. values for C to O and solve for S accordingly

• Inversely scale the components for Poisson noise

Restores normalized count space back to real count space

Rotated PCA Delivers a Spatially Simple Analysis

Rotated PCA

- Based on a Varimax rotation of PCA images
- Relaxes orthogonality of one dimension
 - Separates data based on simplifying images or spectra
 - Works well with data containing large spectral backgrounds (AES, XPS)
- Minor spectral features can be hard to find.

Q. Where is the spatial contrast in the image, and what is it?

Spatially Simple Rotation



How Many MVS Components will Describe My Data?



Eigen Number

Keenan-Kotula assume no more than 25 components



Paths to Phase Analysis



ion into

Segment chemical

Quant

Compute MVS spatial simplicity solution (mathematical result)

Compute quantified X-ray images usually spatially binned

Perform X Phase[†] Perform X Phase[†]

Quantify aggregate pixels for each phase

Quantify aggregate pixels for each phase

⁺X Phase output: Binary masks showing location of phases and associated spectra (real data)

Phase Segmentation: X Phase



Input: Compass (Spectral/Spatial simplicity) results, or, X-ray images (raw cts. or quantified)

- Define min area for a phase (default = 1.5%)
- Perform histogram stretch on each image
- Multiply each image by the others, and stretch resulting 16 bit image
- Threshold image and binarize
- If mask is >1.5% of tot. area, declare it a phase

More Phase Segmentation X Phase

 If overlap occurs then combine overlap into the first detected phase for that area

 From each binary mask extract a spectrum from the datacube

Now you have a image (showing the location) and spectrum (giving the composition) pair for each chemical compound, or phase.

• Compare phase spectra to a library of stored phases using conventional least squares fit and give it a name Note: Smithsonian mineral library will be available soon-

Quantification: Instrumental Conditions



Quantification: "Standardless" EDS: *Useful or Worthless?*



Quantification: Standards-based EDS: K-hornblende as primary std



Parícutin Volcano

Mineralogical-Geological Glossary

| Mineral name | End-member formulae |
|---------------|-------------------------|
| plagioclase | NaAlSizOz - CaAlzSizOz |
| pragroenaee | |
| olivine | $Mg_2SiO_4 - Fe_2SiO_4$ |
| orthopyroxene | $MgSiO_3 - FeSiO_3$ |

| term | definition | | |
|--------|---|--|--|
| scoria | Rapidly cooled vesicular volcanic rock, darker than pumice. | | |
| ash | Particles of crystals and glass ejected into the atmosphere < 2 mm in diameter. | | |



The Extraordinary Story of Parícutin Volcano

- In Feb. 1943 the volcano was born in the presence of several Tarascan indians, erupted for 9 years
- RARE event!
- Lava flows covered 25 km² burying 2 villages, killing 4500 cattle, 550 horses, and forcing the evacuation of several thousand people.
- Smithsonian has a special connection to the eruption as William Foshag, curator of mineralogy was in Mexico studying ore deposits when the birth occurred and documented the first 2.5 years of the volcano's life
- Foshag was an avid photographer and took lots of 16 mm footage

Play video with Overly Dramatic Soundtrack: 1943 Newsreel Play video of man rolling a recently solidified rock off lava flow... used to cook dinner by Smithsonian scientist



The Specimens

Scoria





Ash

108135

116301-4

Note nearly bi-modal distribution of black and brown particles...magma mixing? Subtle compositional differences?

Scoriae Particles





------- 500 μm -------Paricutin 116301-4 ash

 WD
 HFW
 det
 HV
 mag
 spot

 5.5 mm
 799 μm
 BSED
 12.50 kV
 378 x
 5.0

· 300 µm ·

Ash Particles



Complicating factors

Geometry makes for variable TOA in unfilled vesicles and hence spectral, & later phase, artifacts

 WD
 HFW
 det
 HV
 mag
 spot

 5.4 mm
 133 μm
 BSED
 12.50 kV
 2 263 x
 5.0

Paricutin 108

Microstructural complexity seemingly fractal in dark particles

WD HFW det HV mag spot <u>—— 1 μm</u>—— 5.5 mm 4.64 μm BSED 12.50 kV 65 000 x 4.0 Paricutin 116301-4 ash





X Phase Results:



keV

108135scoria1(2) glass Full scale counts: 509771 Composite 800000 glass 600000 phase overlay 200000 onto BSE Ca Ti Ti Cr ull scale counts: 1872162 108135scoria1(2) epoxy-void 3000000 These epoxy + void 2500000 000000 500000 000000 genuine images are Fe Na Mg AL SI PP S S CI CI K K Ca Ti Ti 500000 Fe Fe binary data olivine 108135scoria1(2) olivine Full scale counts: 48397 masks, eg 60000 olivine the phase is either keV Full scale counts: 509111 108135scoria1(2) plagioclase there or B00000 plagioclase not Ca K K Ca Ti Ti Cr Fe CI CI



Area = 22.7%

Glass



| Element | Weight % | Atom % | Compnd % |
|---------|----------|--------|----------|
| Line | | | |
| O K | 46.4S | 62.4 | |
| Na K | 3.3 | 3.1 | 4.4 |
| Mg K | 1.7 | 1.5 | 2.9 |
| Al K | 7.1 | 5.7 | 13.5 |
| Si K | 27.4 | 20.9 | 58.6 |
| P K | 0.4 | 0.3 | 0.9 |
| S K | 0.3 | 0.2 | 0.8 |
| KK | 1.2 | 0.6 | 1.4 |
| Ca K | 3.2 | 1.7 | 4.5 |
| Ti K | 1.1 | 0.5 | 1.9 |
| Cr K | 0.1 | 0.0 | 0.1 |
| Mn K | 0.2 | 0.1 | 0.2 |
| Fe K | 7.5 | 2.9 | 10.8 |
| Total | 100.0 | 100.0 | 100.0 |

| Element | Weight % | Atom % |
|---------|----------|--------|
| Line | | |
| O K | 46.8 | 62.7 |
| Na K | 3.3 | 3.1 |
| Mg K | 1.7 | 1.5 |
| Al K | 7.1 | 5.6 |
| Si K | 27.2 | 20.8 |
| P K | 0.4 | 0.3 |
| S K | 0.3 | 0.2 |
| KK | 1.2 | 0.6 |
| Ca K | 3.2 | 1.7 |
| Ti K | 1.1 | 0.5 |
| Cr K | 0.1 | 0.0 |
| Mn K | 0.2 | 0.1 |
| Fe K | 7.5 | 2.9 |
| Total | 100.0 | 100.0 |



Area = 50.9%

Epoxy + Void



Not a POI



Area = 2.4%

Olivine



| Element | Weight % | Atom % | Compnd % | Elei |
|---------|----------|--------|----------|------------|
| Line | | | | Li |
| O K | 43.2S | 59.6 | | 0 |
| Na K | 0.8 | 0.8 | 1.1 | Na |
| Mg K | 17.1 | 15.5 | 28.4 | M |
| Al K | 2.0 | 1.6 | 3.7 | Al |
| Si K | 19.5 | 15.3 | 41.8 | Si |
| P K | 0.1 | 0.1 | 0.3 | P |
| S K | 0.1 | 0.1 | 0.4 | S |
| KK | 0.2 | 0.1 | 0.3 | K |
| Ca K | 1.0 | 0.6 | 1.4 | Ca |
| Ti K | 0.3 | 0.1 | 0.5 | Ti |
| Mn K | 0.4 | 0.2 | 0.5 | M |
| Fe K | 15.2 | 6.0 | 21.7 | Fe |
| Total | 100.0 | 100.0 | 100.0 | Tot |

| Element | Weight % | Atom % |
|---------|----------|--------|
| Line | | |
| O K | 40.4 | 56.7 |
| Na K | 0.9 | 0.8 |
| Mg K | 17.9 | 16.5 |
| Al K | 2.1 | 1.7 |
| Si K | 20.6 | 16.5 |
| P K | 0.1 | 0.1 |
| S K | 0.1 | 0.1 |
| KK | 0.2 | 0.1 |
| Ca K | 1.1 | 0.6 |
| Ti K | 0.3 | 0.1 |
| Mn K | 0.4 | 0.2 |
| Fe K | 15.9 | 6.4 |
| Total | 100.0 | 100.0 |



Area = 24.0%

Plagioclase



| Element | Weight % | Atom % | Compnd % | Element | Weight % | Atom % |
|---------|----------|--------|----------|---------|----------|--------|
| Line | | | | Line | | |
| O K | 46.7S | 61.7 | | O K | 44.7 | 59.8 |
| Na K | 3.0 | 2.8 | 4.1 | Na K | 3.1 | 2.9 |
| Mg K | 0.8 | 0.7 | 1.3 | Mg K | 0.8 | 0.7 |
| Al K | 13.2 | 10.3 | 24.9 | Al K | 13.6 | 10.8 |
| Si K | 24.8 | 18.7 | 53.1 | Si K | 25.8 | 19.7 |
| P K | 0.2 | 0.1 | 0.5 | РК | 0.2 | 0.2 |
| S K | 0.2 | 0.2 | 0.6 | S K | 0.2 | 0.2 |
| Cl K | 0.0 | 0.0 | 0.0 | Cl K | 0.0 | 0.0 |
| KK | 0.4 | 0.2 | 0.4 | KK | 0.4 | 0.2 |
| Ca K | 7.9 | 4.2 | 11.1 | Ca K | 8.2 | 4.4 |
| Ti K | 0.3 | 0.1 | 0.5 | Ti K | 0.3 | 0.1 |
| Cr K | 0.1 | 0.0 | 0.1 | Cr K | 0.1 | 0.0 |
| Fe K | 2.5 | 0.9 | 3.5 | Fe K | 2.6 | 1.0 |
| Total | 100.0 | 100.0 | 100.0 | Total | 100.0 | 100.0 |