

Physics 353: Problem Set 3 - SOLUTIONS

1 Kittel and Kroemer Chp. 4 #18. Isentropic expansion of a photon gas.

$$(a) \quad T_i V_i^{1/3} = T_f V_f^{1/3} \quad V \propto T^3$$

$$T_f = 2.7 \text{ K} \quad T_i = 3000 \text{ K}$$

$$\therefore T_i V_i = T_f V_f$$

$$\therefore \frac{V_f}{V_i} = \frac{T_i}{T_f} = \frac{3000 \text{ K}}{2.7 \text{ K}} = 1111$$

$$(b) \quad dQ = 0$$

$$\therefore dW = -dU \quad W = -\int dU$$

$$\therefore W = -(U_{\text{final}} - U_{\text{initial}})$$

$$= \frac{\pi^2}{15 c^3 k^3} (V_i T_{\text{initial}}^4 - V_f T_{\text{final}}^4)$$

$$\left. \begin{array}{l} T_i V_i^{1/3} = T_f V_f^{1/3} \\ T_i^3 V_i = T_f^3 V_f \end{array} \right) = \frac{\pi^2}{15 c^3 k^3} V_i T_i^3 (T_i - T_f)$$

2 Phonon heat capacity.

$$U = V \frac{3\hbar}{2\pi^2 c_s^3} \int_0^{\omega_{\max}} \frac{\omega^3 d\omega}{e^{\hbar\omega/\tau} - 1}$$

Low τ . " $\omega_{\max} \approx \infty$ ". $x \equiv \frac{\hbar\omega}{\tau}$.

$$U \approx V \frac{3\hbar}{2\pi^2 c_s^3} \left(\frac{\tau}{\hbar}\right)^4 \int_0^{\infty} \frac{x^3 dx}{e^x - 1}$$

$$c_s^3 = \frac{\omega_{\max}^3 V}{6\pi^2 N} \text{ (class)} \quad \downarrow \text{integral} = \frac{\pi^4}{15} \text{ (class)}$$

$$\rightarrow U \approx \cancel{V} \frac{3\tau^4}{2\pi^2 \hbar^3} \cancel{\omega_{\max}^3 V} \frac{6\pi^2 N}{15}$$

Therefore $c_v = \left. \frac{\partial U}{\partial \tau} \right|_V = \frac{12}{5} \pi^4 N \left(\frac{\tau}{\Theta_D} \right)^3$, simplifying the algebra and using the definition of the Debye temperature $\Theta_D \equiv \hbar\omega_{\max}$.

2b

(b) High τ . $\exp\left(\frac{\hbar\omega}{\tau}\right) \approx 1 + \frac{\hbar\omega}{\tau}$, $\frac{\omega^3}{e^{\hbar\omega/\tau} - 1} \approx \frac{\omega^3}{\hbar\omega/\tau} = \frac{\omega^2 \tau}{\hbar}$

$$\text{So } U \approx V \frac{3\hbar}{2\pi^2 c_s^3} \int_0^{\omega_{\max}} \frac{\omega^2 \tau}{\hbar} d\omega = \frac{V \cancel{3\hbar} \tau}{2\pi^2 c_s^3 \cancel{\hbar}} \frac{\omega_{\max}^3}{3}$$

$$\text{From class } \omega_{\max} = (6\pi^2 c_s^3 N/V)^{1/3},$$

$$\text{So } U \approx \frac{\cancel{3} V}{2\pi^2 c_s^3} \tau \frac{6\pi^2 c_s^3 N}{V} = 3N\tau.$$

$$c_v = \left. \frac{\partial U}{\partial \tau} \right|_V = \frac{\partial U}{\partial \tau} \Big|_V = 3N \quad \checkmark$$

(since PdV=0)

3 Photon Energy.

a. isothermal stage. $\delta Q = \delta U + \delta W = \delta U + P \delta V$.

$$\delta Q = \left(\frac{\partial U}{\partial z} dz + \frac{\partial U}{\partial V} dV \right) + P dV \quad (\text{general})$$

$$\delta Q = \frac{\partial U}{\partial V} dV + P dV \quad (\text{isothermal stage})$$

b. Work done over the entire cycle $\delta W = dP dV$.

Heat = heat input in part (a). $\rightarrow \eta = \frac{W}{Q_H}$

$$= \eta = \frac{dP dV}{\frac{\partial U}{\partial V} dV + P dV} = \frac{dP}{\frac{\partial U}{\partial V} + P}$$

$$\text{Carnot: } \eta = 1 - \frac{T_C}{T_H} \quad (\text{ideal}) = \frac{dT}{T+dT}$$

$$\Rightarrow \frac{W}{Q_H} = \frac{dT}{T+dT}$$

$$\text{c. } \dots \Rightarrow \frac{dP}{\frac{\partial U}{\partial V} + P} = \frac{dT}{T+dT} \Rightarrow dP (T+dT) = dT \left(\frac{\partial U}{\partial V} + P \right)$$

Let's now use the fact that $P = aT^4$. $\Rightarrow \frac{dP}{dT} = 4aT^3$.

$$\Rightarrow 4aT^3 (T+dT) = \frac{\partial U}{\partial V} + aT^4$$

$$\Rightarrow 4aT^4 + 4aT^3 dT = aT^4 + \frac{\partial U}{\partial V}$$

$$\Rightarrow 3aT^4 + 4aT^3 dT = \frac{\partial U}{\partial V}$$

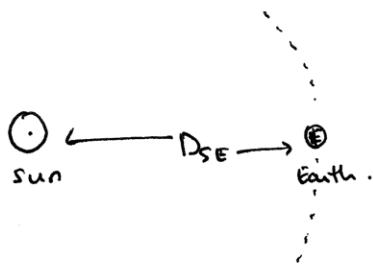
no differentials a differential a derivative.

In the limit of small differentials, the second term $\rightarrow 0$.

$$\Rightarrow 3aT^4 = \frac{\partial U}{\partial V} \Rightarrow \underline{U = 3aV T^4}$$

(assuming $U \rightarrow 0$ as $V \rightarrow 0$).

4 Surface temperature of the Earth – Part 1. Kittel and Kroemer Chp. 4 #5.



Total power radiated by the sun = $J_u \times \text{sun's surface area}$
 $= P_0 = \sigma_B T_0^4 4\pi R_0^2$

Total solar power received by earth = $P'_0 = P_0 \frac{\text{area of earth}}{\text{area of sphere at } D_{SE}}$
 $\rightarrow P'_0 = P_0 \frac{\pi R_E^2 \leftarrow \text{disc}}{4\pi D_{SE}^2 \leftarrow \text{sphere}}$

Total power radiated by the earth = $J_u \times \text{earth's surface area}$
 $= P_E = \sigma_B T_E^4 4\pi R_E^2$

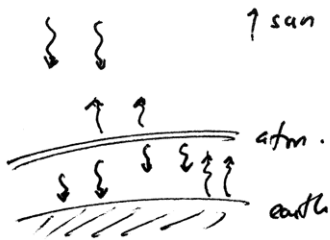
Equil: $P_E = P'_0 \rightarrow \sigma_B T_0^4 4\pi R_0^2 \frac{\pi R_E^2}{4\pi D_{SE}^2} = \sigma_B T_E^4 4\pi R_E^2$

$\rightarrow \frac{T_0^4 R_0^2}{D_{SE}^2} = T_E^4 \rightarrow T_E = \left(\frac{T_0^4 R_0^2}{4 D_{SE}^2} \right)^{1/4}$

or $T_E = T_0 \left(\frac{R_0}{2 D_{SE}} \right)^{1/2}$

Numerically: $T_E = 280 \text{ K} = 7^\circ \text{C}$. a bit cold...

5 Surface temperature of the Earth – Part 2. The greenhouse effect.



Earth is in radiative equilibrium w/ atm & sun -

$$\text{i.e. } P_E \text{ (see problem 4)} = P'_0 + P_{\text{atm}} \frac{1}{2}$$

↑
radiated by atmosphere,
downward.

Atmosphere is in radiative equil. w/ earth. (absorbs none of sun's radiation).

$$P_{\text{atm}} = P_E$$

$$\rightarrow P_E = P'_0 + \frac{1}{2} P_E \rightarrow P_E = 2P'_0$$

$$\rightarrow \sigma_B T_E^4 4\pi R_E^2 = 2 \sigma_B T_0^4 4\pi R_0^2 \frac{\pi R_E^2}{4\pi D_{SE}^2} \quad (\text{see \# 4})$$

$$\rightarrow T_E^4 = 2 \left(\frac{T_0^4 R_0^2}{4 D_{SE}^2} \right) \rightarrow T_E = T_0 \underbrace{2^{1/4} \left(\frac{R_0}{2 D_{SE}} \right)^{1/2}}$$

i.e. the T from #4 times $2^{1/4}$.

$$\text{Numbers: } \underline{T_E = 333\text{K} = 60^\circ\text{C}} \quad \text{A bit warm!}$$

6 Kittel and Kroemer Chp. 4 #8.

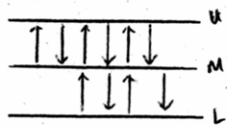


Plate M must be in radiative equilibrium with plates U and L. The radiative flux density hitting a unit area of it (considering both sides) must equal the radiative flux density of its emission (considering both sides).

$$J_{UM} + J_{LM} \begin{array}{l} \text{arrows to the} \\ \text{middle plate} \end{array} = J_{MU} + J_{ML} \begin{array}{l} \text{arrows from the} \\ \text{middle plate} \end{array}$$

$$\sigma T_U^4 + \sigma T_L^4 = \sigma T_M^4 + \sigma T_M^4 = 2\sigma T_M^4 \Rightarrow T_M = \left[\frac{1}{2} (T_U^4 + T_L^4) \right]^{1/4}$$

Energy flux density without the middle plate = $J_0 = \sigma (T_U^4 - T_L^4)$

Considering the lower half of the drawing, the energy flux density in the vacuum between plates M and L = J_{low}

$$= \sigma (T_M^4 - T_L^4) = \sigma \left(\frac{1}{2} (T_U^4 + T_L^4) - T_L^4 \right) = \sigma \left(\frac{1}{2} (T_U^4 - T_L^4) \right)$$

$$J_{low} = \frac{1}{2} J_0$$

Similarly considering the upper half, $J_{up} = \sigma (T_U^4 - T_M^4)$

$$= \sigma \left(T_U^4 - \frac{1}{2} T_U^4 - \frac{1}{2} T_L^4 \right) = \sigma \left(\frac{1}{2} (T_U^4 - T_L^4) \right) \Rightarrow J_{up} = \frac{1}{2} J_0$$

7 Chemical Potential of an Ideal Gas.

$$z_{i, PB} = n_Q V \quad \dots \quad z = \frac{(z_i)^N}{N!} \quad z_i = z_{i, PB} z_{int}$$

$$\begin{aligned} \text{a. } F &= -\tau \ln z \Rightarrow F = -\tau \ln \frac{z_i^N}{N!} = -\tau \ln z_i^N + \tau \ln N! \\ &= F = -N\tau \ln z_i + \tau (N \ln N - N) \quad (\text{Stirling}) \\ &= F = -N\tau \ln z_{i, PB} - N\tau \ln z_{int} + \tau (N \ln N - N). \end{aligned}$$

$$\mu = \left. \frac{\partial F}{\partial N} \right|_{\tau, V}, \text{ so } \mu = -\tau \ln z_{i, PB} - \tau \ln z_{int} + \tau \left(N \frac{1}{N} + \ln N - 1 \right)$$

$$= \mu = -\tau \ln(n_Q V) - \tau \ln z_{int} + \tau \ln N$$

$$= \mu = \tau \ln \left(\frac{N}{n_Q V z_{int}} \right); \quad n \equiv \frac{N}{V}, \text{ so}$$

$$\mu = \tau \ln \left(\frac{n}{n_Q z_{int}} \right)$$

b. It is true that $U = \frac{3}{2} N\tau$, and $\mu = \left. \frac{\partial U}{\partial N} \right|_{V, \sigma}$. However we can't simply evaluate μ by differentiating U since this partial derivative **must** be taken at constant entropy and volume. It is not at all evident from $U = \frac{3}{2} N\tau$ how U would change as a function of N if entropy and volume were held fixed. Entropy for example varies in a complicated way as a function of N , τ , etc. — varying N but keeping entropy fixed, we would certainly have to change τ , which would affect our derivative.